S1 Table. Comparison of volumetric energy densities for fused cyclobutanes and petroleum-derived and renewable jet fuels

	Volumetric energy
Compound or fuel	density (MJ/L)
Cyclobutane, C <sub>4</sub> H <sub>8</sub>	$33.0^{a}$
[2]-ladderane, $C_6H_{10}$	42.3 <sup>a</sup>
[3]-ladderane, $C_8H_{12}$	46.3 <sup>a</sup>
[4]-ladderane, $C_{10}H_{14}$	49.2 <sup>a</sup>
[5]-ladderane, C <sub>12</sub> H <sub>16</sub>	51.4 <sup>a</sup>
Jet fuel (kerosene) <sup>b</sup>	35.06°
Bio-SPK <sup>b</sup>	33.2 <sup>b</sup>

<sup>&</sup>lt;sup>a</sup> Calculated using entropies of formation (Novak 2008) and estimated densities from ACD/Labs' PhysChem Suite software

Hemighaus, G., et al. (2004) Aviation Fuels Technical Review, Chevron Corporation.

Kallio, P., et al. (2014) Renewable jet fuel, Current Opinion in Biotechnology 26, 50-55

Novak, I. (2008) Ring strain in [n]ladderanes, J. Phys. Chem. A 112, 10059-10063.

<sup>&</sup>lt;sup>b</sup> This fuel is the most relevant to commercial aviation in the U.S. Bio-SPK, Bio-Derived Synthetic Paraffinic Kerosene, is a renewable jet fuel in use today (Kallio et al. 2014)

<sup>&</sup>lt;sup>c</sup> From Hemighaus et al. (2004)